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Contribution to the study of the structural and dielectric properties of magnesium doped BaTiO₃ material

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Abstract

In this work the different materials of composition Ba_{1-x}Mg_xTiO₃ (BMxT) (x between 0 and 1) powders were successfully synthesized by the sol-gel method and their structure was characterized by X-ray diffraction. Scanning electron microscopy (SEM) investigation showed an increase in grain size with increasing of Mg content and calcining temperature. Dielectric measurements indicated that when the rate of Mg increases the temperature T_m decreases to the minimum value of 105 ° C around 15% in Magnesium.

Keywords: Ceramic, sol-gel, Dielectric properties, Doping, BMxT.

Introduction

Perovskite-type ferroelectric materials are characterized by a very high dependence of some of their properties both with the electric field, the mechanical and thermal stresses applied and also with the procedure for their production. BaTiO₃ has been extensively used in the electronics industry due to its high dielectric constant and low loss characteristics [1–5].

Barium titanate has been emerged as a new material, which typically undergo a marked relaxation in dielectrics properties and induced to a decrease in relative permittivity by frequency, with a resonance peak in the dielectric loss at a narrow frequency band. BaTiO₃ has been attracting attention due to its particular multilayer capacitors, optical devices, thermistors and electromechanical properties [6].

BaTiO₃ can also exist in the orthorhombic phase at a temperature of 0°C, in the rhombohedral phase below -90°C, BaTiO₃ exists in the hexagonal phase above 1460°C. Although the primitive cube is the idealized structure, the differences in radius between cations A and B can modify the structure at several different so-called distortions, whose inclination is most frequent. With the inclination of perovskite, the octahedron BO₆ twists. Along one or more

axes to accommodate the difference. The ferroelectric properties of ABO_3 type ceramic perovskite (here barium titanate) can be effectively controlled by doping using different doping elements [7-9].

Several synthesis methods have been used to prepare BT powders including nonconventional ones such as, oxalate, hydrothermal synthesis, citrates and polymeric precursors method, mainly based on the Pechini-type process and the sol-gel method [10-11].

The sol-gel method offers numerous advantages such as an excellent control of the stoichiometry, a good homogeneity of the powders and low processing temperature [12].

Experimental Procedure

The synthesis of various compounds of Barium titanate, doped with magnesium $Ba_{1-x}Mg_xTiO_3$ (BMxT) corresponding to substitutions the Mg^{2+} by Ba^{2+} was performed by sol-gel method.

The $Ba_{1-x}Mg_xTiO_3$ materials are prepared from solutions of magnesium acetate, barium acetate and titanium sol, mixed in stoichiometric proportions. The BMxT solutions obtained are dried in an oven at $80^\circ C$ for 72 hours leading to the formation of a gel. The latter is dried, crushed and then calcined at $1000^\circ C$ for 4 hours.

Phase identification of the samples was performed using X-ray diffraction (Cu K ray, $\lambda = 1.5418 \text{ \AA}$), scanning electron microscopy and Dielectric spectrum.

RESULTS AND DISCUSSION

The XRD analysis (Fig. 1) revealed that BMxT ceramic samples (for $x = 0, 0.1, 0.2, 0.3, 0.4$ and 1) sintered at $1000^\circ C$ for 4 hours, shows the appearance of three distinct domains. The first domain ranges from $x=0$ to 0.15 crystallized into a single perovskite structure without changing the symmetry of the structure $BaTiO_3$. The second domain ranges from $x \geq 0.2$, all the diffraction peaks, indicating the pseudocubic BMxT appeared in the patterns but with presence of the two secondary phases MgO and $BaMg_6Ti_6O_{19}$ and the last domain for $x = 1$, the characteristic peaks of pure $MgTiO_3$ perovskite phase are observed [13].

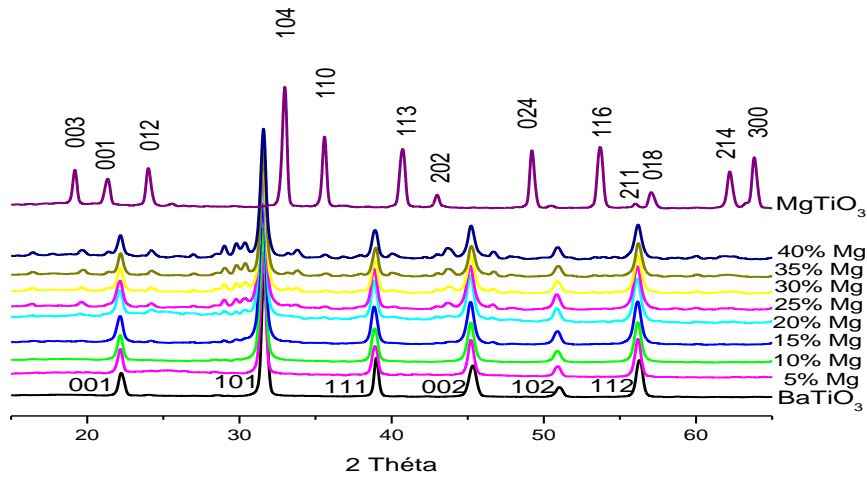


Figure1: XRD spectra of the $Ba_{1-x}Mg_xTiO_3$ phases heat treated at $1000^\circ C$ for 4h

Characterization by scanning electron microscopy

The SEM micrographs of ceramic samples (BMxT) sintered at $1100^\circ C$ for 8h are shown in Fig.2. Show the morphology and appearance of grains of BMxT ceramic ($x = 0, 0.1, 0.2, 0.3$ and 0.4), they show an almost homogeneous structure with grains of different sizes and low porosity.

It is observed that the size of the grains progressively increases with the concentration of Mg. Indeed, the latter diffuses from the interfaces between the grains in the joints, thus favoring the enlargement of the grains with the increase magnesium rate in materials of composition $Ba_{1-x}Mg_xTiO_3$

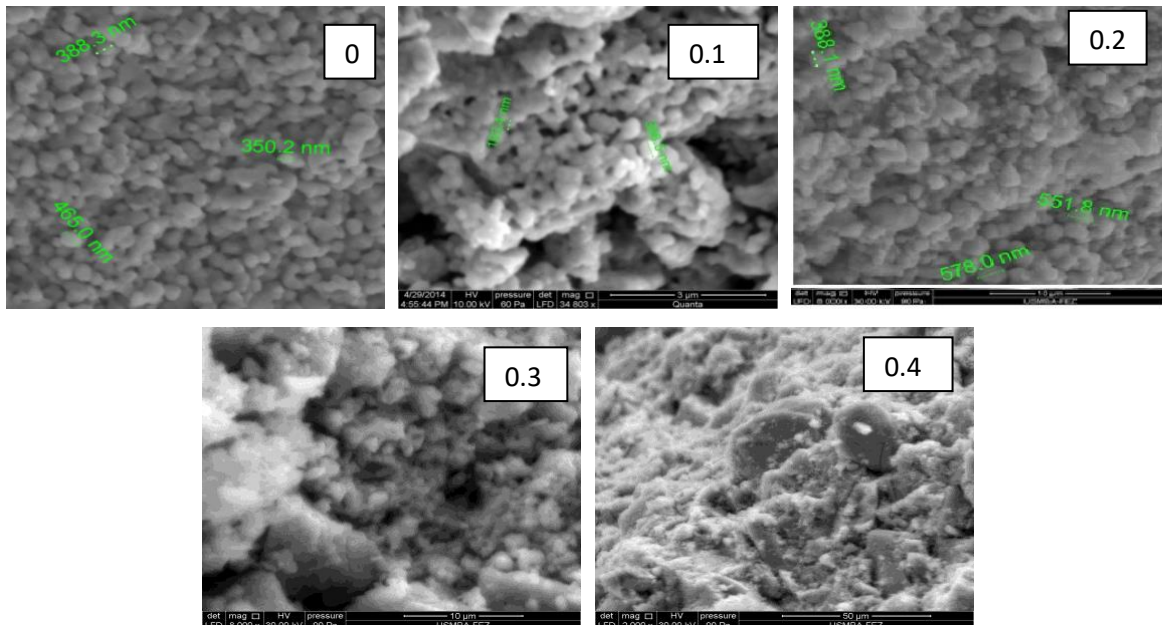


Figure 2. SEM micrographs of the $Ba_{1-x}Mg_xTiO_3$ ($x=0, 0.1, 0.2, 0.3$ and 0.4) ceramic sintered at $1100^\circ C$ for 8h

The average grain size for the sample BMT(0) (Fig. 2) varies from approximately 350.2 nm to 465 nm and from approximately 388.1 and 578 nm from the samples BMT(1) and BMT(2), respectively (Fig. 2), which behavior is in accordance with that observed on XRD patterns.

Dielectric properties

Variation of the dielectric permittivity (ϵ_r) as function of the temperature, for different frequencies, are presented in figure 4, which shows that the relative permittivity ϵ_r increases when the temperature increases and passes through a maximum ϵ_{rmax} diffused at the temperature T_m , then decreases. The shape of the variation of ϵ_r at temperature T_m is independent of the frequency, this suggests that it is a classic transition.

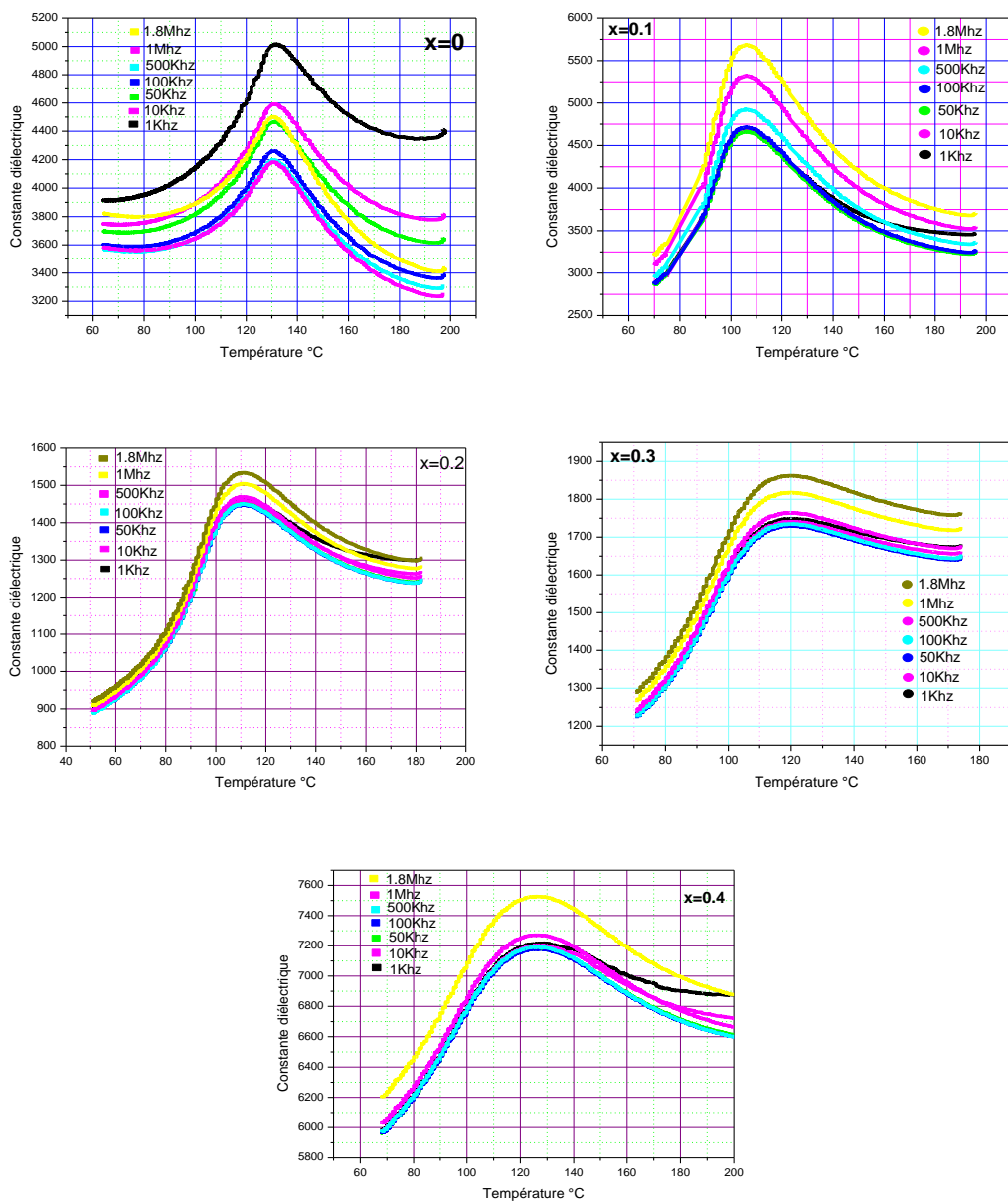


Figure. 3. Thermal variations of the relative permittivity, ϵ_r of BMxT ($x=0, 0.1, 0.2, 0.3, 0.4$) heat treated at 1100°C/8H

It is observed (Table 1), that the insertion of Mg causes a decrease in the temperature T_m . From the value 130°C for BaTiO₃ ($x = 0$) up to 105°C for a rate of 10%. On the other hand, the permittivity ϵ_{max} undergoes a slight increase (from 5000 for $x = 0$ to 5300 for $x = 0.1$).

Above 10% Mg, there is a gradual increase in temperature T_m up to 125 ° C. On the other hand, the value of the dielectric permittivity ϵ_r undergoes a rapid fall for $x = 0.2$. ($\epsilon_{max} = 1700$), Then for $x= 0.4$, the maximum dielectric permittivity ϵ_r increases and reaches higher values (7300). This variation indicates that Mg is inserted into the crystallographic site A of coordination 12 of the perovskite BaTiO₃, by substituting the Ba. Indeed this phenomenon is comparable to that observed during the BT study doped with Strontium [14].

x (%) Magnésium	T_m (°C)	ϵ_{max}
0	130	5000
10	105	5300
20	110	1700
30	120	1850
40	125	7300

Table1. Comparison of ϵ_r and T_m corresponding to samples of BMxT

Dielectric losses:

Table 2 groups the dielectric losses of BMxT compounds as a function of the Mg rate for a frequency of 1.8 KHz.

It should be noted that the dielectric losses at ambient temperature are very low compared to that corresponding to pure BT (0.1272) to 0.0124 up to 0.0082 for $x = 0.1$ and 0.2.

This reduction in dielectric losses compared to that observed for the pure BaTiO₃ compound, may be due to the insertion of Mg, known by its effect of reducing dielectric losses, in the BaTiO₃ lattice [15].

Taux de Mg (%)	BT (x=0)	10%	20%	30%	40%
Dielectric losses at T_m	0.1272	0.0124	0.0082	0.0142	0.0103

Table.2. Dielectric losses of BMxT

Diffuse phase transition:

The diffuse phase transition is a transformation of the ferroelectric phase to the paraelectric phase around the maximum temperature (T_m), accompanied by a frequency dispersion above T_m [16]. Figure 4 shows the fitting simulation of ϵ_r by the modified Uchino' law, depending on the temperature of the different compounds BMxT at a frequency of 1.8 KHz.

$$\frac{1}{\varepsilon} = \frac{1}{\varepsilon_m} \left[1 + \frac{(T-T_m)^\gamma}{2\delta^\gamma} \right] \quad [17].$$

γ and δ are the relaxor and diffuse parameters respectively.

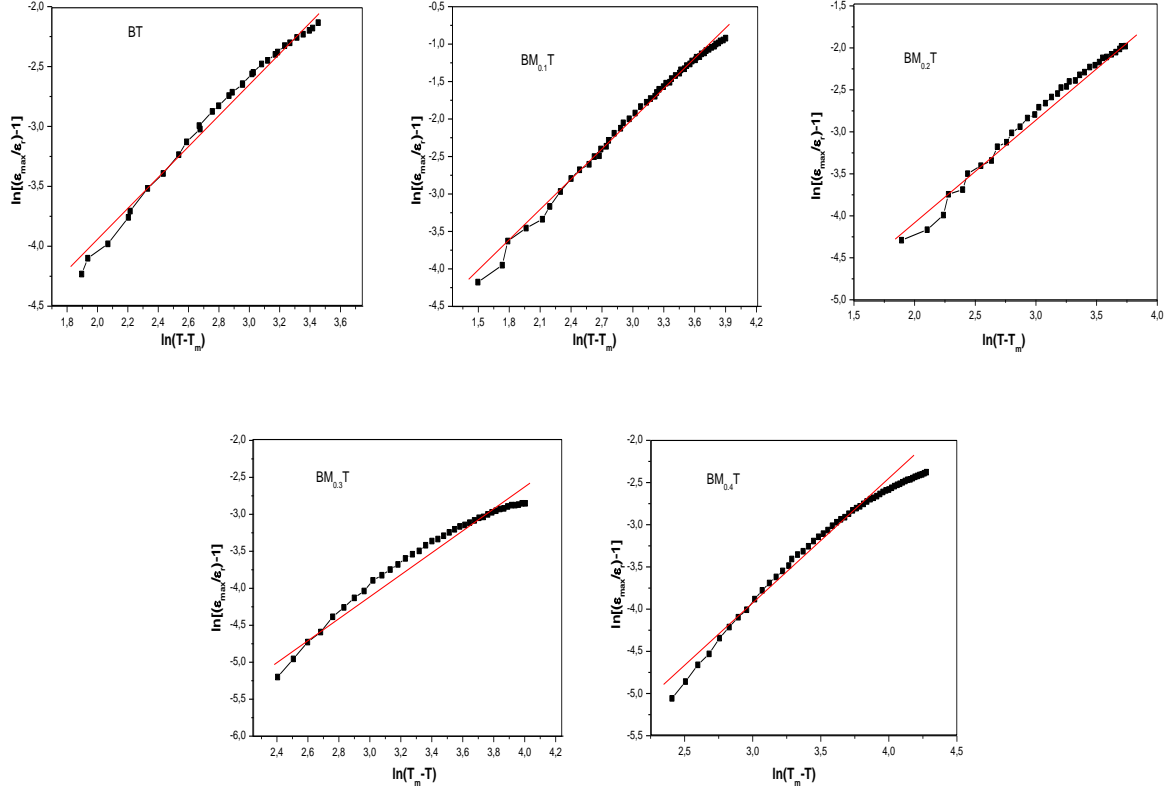


Figure.4. Plot of $\ln(\varepsilon_m/\varepsilon - 1)$ as a function $\ln(T_m-T)$

Magnesium (%)	γ	δ
0	1,287	22.46
10	1,409	39,84
20	1,575	48,41
30	1,588	55,43
40	1,721	63,11

Table-3. Parameter values (γ , δ) for BMxT for frequency 1.8 KHz

γ and δ values obtained are given in Table-3. The values show that these constants increase with magnesium concentration and the transition shows a diffuse character ($1 < \gamma < 2$).

Conclusion

In summary, the study of the effect of substitution of Ba by Mg on the physicochemical and dielectric properties of the compounds $Ba_{1-x}Mg_xTiO_3$ produced by the sol-gel method for different concentrations (0, 0.1, 0.2, 0.3 and 0.4) resulted in that Mg has an effect on both the transition temperature, the constant and the dielectric losses. Dielectric measurements revealed a diffuse ferro-to-paraelectric phase and confirmed by the modified Ushino's law. Note that when the Mg level increases, the temperature T_m decreases to the minimum value of $105^\circ C$ around a rate x of 15% in Magnesium. This decrease in temperature T_m indicates that Mg is inserted into the crystallographic site A of the perovskite $BaTiO_3$.

Using scanning electron microscopy on different samples of BM_xT showed that magnesium affects grain shape and size, the more the Mg concentration increases, the more the grain sphericity increases. This reflects a new rearrangement at the level of the mesh.

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